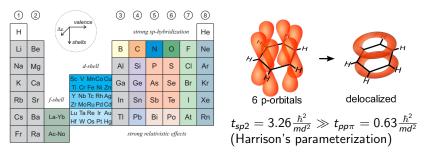
Intro 1/4: What is organic semiconductor

- Organic built from C with terminal H, possibly with isovalent substitutions (N for CH, O/S for CH₂, F/Cl for H)
- 2. π -conjugated only π -electrons on frontier orbitals

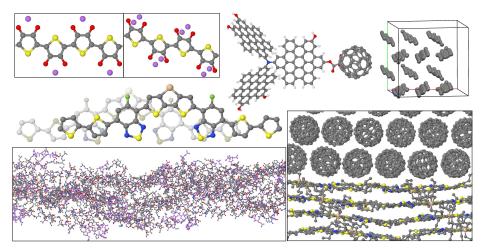


- Also metal-organic include d-elements via pdπ
- Also hybrid include any electronic system in resonance

Number of organic semiconductors is comparable to number of inorganic ones Display is an example of a competitive application

Intro 2/4: Structural diversity

From 0D to 3D, from rigid to soft, often heterogeneous with multiscale structuring



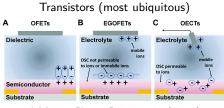
Intro 3/4: Some applications of organic semiconductors

Any electronic device can be made all-organic

Light emitters (most successful)

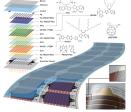


Adv Mater 33, 2005630 (2021)



J Mater Chem C 6, 11778 (2018)

Solar cells (approaching 20% PCE) Adv Energy Mater 11, 2002653 (2021)



Energy Environ Sci 7, 2925 (2014)

Batteries (early research) $\begin{array}{c} & & \\ &$

Chem Rev 120, 6490 (2020)

Intro 4/4: Challenges in modeling organic semiconductors

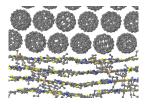
• Electron-electron correlations in low dimensional systems

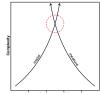
Trans-polyacetylene (tPA) – 5 electrons in repeating unit – smallest real-world material for which majority of mainstream approaches fail



- Strong mode-specific electron-phonon coupling + soft structure
 - tPA Peierls transition (C–C bond stretching mode) and solitons
 - Bond Length Alternation (BLA)
 - polaron formation
 - vibronic progression in spectra

• Scale gap problem – what is modeled is not what is measured





1nm 10 nm Scale 100 nm 1 µm

